



Pergamon

Tetrahedron Letters 41 (2000) 8877–8880

TETRAHEDRON
LETTERS

Enantioselective synthesis of propargylic alcohols by addition of enantiopure cyclopentadienyldialkoxyallyltitanium complexes to acetylenic aldehydes

Samir BouzBouz,^{a,*} Fabienne Pradaux,^a Janine Cossy,^{a,*}
Clotilde Ferroud^b and Annie Falguières^b

^aLaboratoire de Chimie Organique associé au CNRS, ESPCI, 10 rue Vauquelin, 75231 Paris Cedex 05, France

^bLaboratoire de Chimie Organique associé au CNRS, CNAM, 292 rue Saint Martin, 75141 Paris Cedex 03, France

Received 1 August 2000; accepted 15 September 2000

Abstract

The enantioselective synthesis of propargylic alcohols was achieved by enantioselective allyltitanation of acetylenic aldehydes. © 2000 Elsevier Science Ltd. All rights reserved.

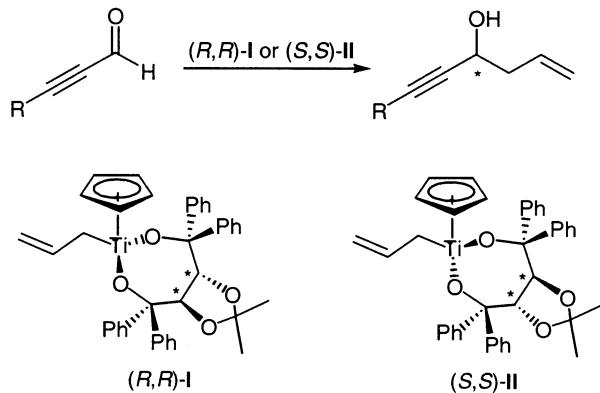
Keywords: allyltitanation; acetylenic aldehydes; propargylic alcohols.

The production of enantiomerically enriched intermediates from achiral starting materials by means of chiral reagents and catalysts has been of increasing interest in recent years.¹ One of the most widely studied goals in this field has been an efficient synthesis of enantioenriched carbinols. Recently, we became interested in the synthesis of chiral secondary propargylic alcohols as these compounds are useful building blocks for the synthesis of biologically² and structurally interesting compounds.³ The methods that have been used for the synthesis of such alcohols involve the enantioselective alkynylation of aldehydes.⁴ Stoichiometric reduction of acetylenic ketones with chirally modified metal hydrides⁵ as well as catalytic asymmetric reduction⁶ or enzymatic reduction⁷ of acetylenic ketones have also been employed to produce these alcohols. Other methods such as cleavage of chiral acetylenic acetals,⁸ enantioselective carbonyl-ene reactions with glyoxylates,⁹ and enantioselective aldol addition reaction to α,β -ynals,¹⁰ in the presence of a chiral catalyst have also found applications.

Additions of chiral allylboranes to α,β -acetylenic aldehydes¹¹ is one of the more general routes to enantioenriched propargylic alcohols. Here, we report a related approach which utilizes the

* Corresponding authors. Fax: 33-1-40-79-46-60; e-mail: janine.cossy@espci.fr

cyclopentadienyldialkoxyallyltitanium complexes (*R,R*)-**I** or (*S,S*)-**II**¹² in such additions. These reagents exhibit remarkably high facial discrimination for such aldehydes despite the relatively small steric requirements of the alkynyl group.



The asymmetric allyltitanation proceeds under neutral conditions at -78°C . Aldehydes **1**¹³ (1 mmol, 1 equiv.) were added to an etheral solution of the cyclopentadienyldialkoxyallyltitanium complex (1.1 equiv.). After 3 h, the reaction mixture was quenched and the propargylic alcohols **2** were isolated in moderately high yield and excellent enantiomeric excess. For each propargylic alcohol, the enantiomeric purity was determined by chiral HPLC.¹⁴ The results are reported in Table 1.

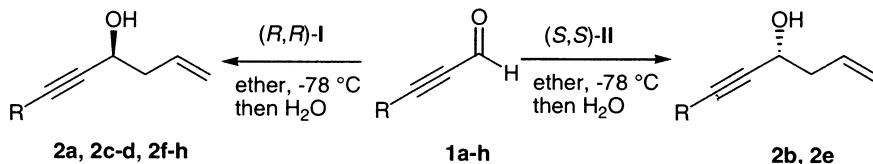


Table 1
Enantioselective allyltitanation of acetylenic aldehydes

Entry	Aldehyde 1	Reagent	Product 2			
			Yield (%)	$[\alpha]_D^d$	Ee (%) ¹⁴	Configuration
a	<i>t</i> -Bu(CH ₃) ₂ Si-	(<i>R,R</i>)- I	53	-28	96	(<i>S</i>) ^a
b	<i>t</i> -Bu(CH ₃) ₂ Si-	(<i>S,S</i>)- II	64	+27	95	(<i>R</i>) ^b
c	Ph-	(<i>R,R</i>)- I	70	-11	93	(<i>S</i>) ^c
d	CH ₃ -(CH ₂) ₆ -	(<i>R,R</i>)- I	71	-19	98	(<i>S</i>) ^a
e	CH ₃ -(CH ₂) ₆ -	(<i>S,S</i>)- II	55	+12	95	(<i>R</i>) ^a
f	TrO-(CH ₂) ₂ -	(<i>R,R</i>)- I	70	-5	93	(<i>S</i>) ^a
g	THPO-CH ₂ -	(<i>R,R</i>)- I	60	-18	95	(<i>S</i>) ^a
h	<i>o</i> -AcO-C ₆ H ₄ -	(<i>R,R</i>)- I	61	-23	95	(<i>S</i>) ^a

^a Determined by ¹H NMR analysis of the *O*-methylmandelates.

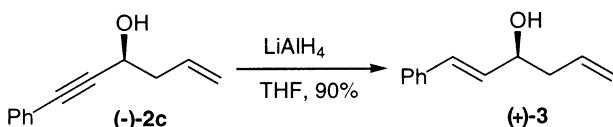
^b Determined by comparison of the $[\alpha]_D$ of (-)-**2a** and (+)-**2b**.

^c Determined by chemical correlation.

^d The optical rotations for compounds **2a-h** were recorded in CHCl₃ at 25°C, with a similar concentration (*c* = 1).

When the propargylic aldehyde **1a** was treated with the (*R,R*)-**I** complex, alcohol (-)-**2a** was isolated in 53% yield (ee = 96%). The absolute configuration was determined by analysis of the ^1H NMR spectra of the corresponding *O*-methylmandelates.¹⁵ The allylic protons of the (*R*)-*O*-methylmandelate appear at higher field than those of the (*S*)-*O*-methylmandelate (0.3 ppm) consistent with the (*S*) configuration for the propargylic alcohol. In a similar manner, when **1c**, **1d**, **1f**, **1g**, and **1h** were treated with the (*R,R*)-**I** complex, the corresponding propargylic alcohols were isolated in 60–70% yield with high enantioselectivity (ee > 93%).

In each case, the absolute configuration was determined as before by analysis of the corresponding *O*-methylmandelates except for alcohol (-)-**2c**. For this compound, the determination of the absolute configuration was achieved by reduction of the propargylic alcohol (-)-**2c** with LiAlH₄ in THF and the vinylcarbinol (+)-**3** was obtained in 90% yield ($[\alpha]_D = +18$, *c* 2, ether; lit.¹⁶: compound (-)-**3**, $[\alpha]_D = -19.3$, *c* 1.33, ether). This chemical correlation allowed us to attribute the (*S*) configuration to compound (-)-**2c**.



Addition of the (*S,S*)-**I** complex to aldehyde **1b** led to alcohol (+)-**2b** in 60% yield (ee = 95%). By comparison of the $[\alpha]_D$ of (-)-**2a** ($[\alpha]_D = -28$) and (+)-**2b** ($[\alpha]_D = +27$), we can assign the (*R*) configuration to (+)-**2b**. When aldehyde **1e** was treated with (*S,S*)-**II**, the propargylic alcohol (+)-**2e** with the (*R*) configuration (determined from the *O*-methylmandelate derivatives),¹⁵ was obtained in 55% yield.

Allyltitanation of α,β -acetylenic aldehydes with the (*R,R*)-**I** and (*S,S*)-**II** complexes proceeds with consistent and predictable stereochemistry to give propargylic alcohols of high optical purity (93–98%). The use of this reaction for the synthesis of biologically active compounds is in progress and will be reported in due course.

Acknowledgements

The authors are indebted to Dr R. O. Duthaler for a generous gift of CpTiCl₃.

References

1. Giacomelli, G.; Lardicci, L. *J. Org. Chem.* **1982**, *47*, 4335–4337.
2. For selected examples, see: [Pheromones]: (a) Mori, K.; Akao, H. *Tetrahedron Lett.* **1978**, 4127–4130. (b) Vigneron, J. P.; Bloy, V. *Tetrahedron Lett.* **1980**, *21*, 1735–1738. [Prostaglandins]: (c) Fried, J.; Sih, J. C. *Tetrahedron Lett.* **1973**, 3899–3902. (d) Kluge, A. F.; Kertesz, D. J.; Yang, C. O.; Wu, H. Y. *J. Org. Chem.* **1987**, *52*, 2860–2868. [Steroids]: (e) Johnson, W. S.; Brinkmeyer, R. S.; Kapoor, V. M.; Yarnell, T. M. *J. Am. Chem. Soc.* **1977**, *99*, 8341–8343. [Alkaloids]: (f) Overman, L. E.; Bell, K. L. *J. Am. Chem. Soc.* **1981**, *103*, 1851–1853. [Palytoxins]: (g) Leder, J.; Fujioka, H.; Kishi, Y. *Tetrahedron Lett.* **1983**, *24*, 1463–1466. (h) Cheon, S. H.; Christ, W. J.; Hawkins, L. D.; Jin, H.; Kishi, Y.; Taniguchi, M. *Tetrahedron Lett.* **1986**, *27*, 4759–4762. [Vitamins E and K]: (i) Cohen, N.; Lopresti, R. J.; Neukom, C.; Saucy, G. *J. Org. Chem.* **1980**, *45*, 582–588. [Roridins and Trichoverrins]: (j) Roush, W. R.; Spada, A. P. *Tetrahedron Lett.* **1982**, *23*, 3773–3776. [Cytochalasins]: (k) Stork, G.; Nakamura, E. *J. Am. Chem. Soc.* **1983**, *105*, 5510–5512. [Serpurene]: (l) Gibbs, R. A.; Okamura, W. H. *J. Am. Chem. Soc.* **1988**, *110*, 4062–4063. [Microcystins]: (m) Kim, H. Y.; Stein, K.; Toogood,

- P. L. J. Chem. Soc., Chem. Commun. **1996**, 1683–1684. [Methylenolactocin]; (n) Zhu, G.; Lu, X. J. Org. Chem. **1995**, *60*, 1087–1089.
3. (a) Colas, Y.; Cazes, B.; Goré, J. *Tetrahedron Lett.* **1984**, *25*, 845–848. (b) Corey, E. J.; Boaz, N. W. *Tetrahedron Lett.* **1984**, *25*, 3055–3058. (c) Henderson, M. A.; Heathcock, C. H. *J. Org. Chem.* **1988**, *53*, 4736–4745. (d) Burger, A.; Hetru, C.; Luu, B. *Synthesis* **1989**, *2*, 93–97. (e) Elsevier, C. J.; Vermeer, P. *J. Org. Chem.* **1989**, *54*, 3726–3730. (f) Marshall, J. A.; Wang, X. *J. J. Org. Chem.* **1992**, *57*, 1242–1252. (g) Myers, A. G.; Zheng, B. *J. Am. Chem. Soc.* **1996**, *118*, 4492–4493. (h) Mukai, C.; Kataoka, O.; Hanaoka, M. *J. Org. Chem.* **1993**, *58*, 2946–2952. (i) Botta, M.; Summa, V.; Corelli, F.; Pietro, G. D.; Lombardi, P. *Tetrahedron: Asymmetry* **1996**, *7*, 1263–1266.
 4. (a) Mukaiyama, T.; Suzuki, K.; Soai, K.; Sato, T. *Chem. Lett.* **1979**, 447–448. (b) Mukaiyama, T.; Suzuki, K. *Chem. Lett.* **1980**, 255–256. (c) Tombo, G. M. R.; Didier, E.; Loubinoux, B. *Synlett* **1990**, 547–548. (d) Niwa, S.; Soai, K. *J. Chem. Soc., Perkin Trans. 1* **1990**, 937–943. (e) Corey, E. J.; Cimprich, K. A. *J. Am. Chem. Soc.* **1994**, *116*, 3151–3152.
 5. (a) Brinkmeyer, R. S.; Kapoor, V. M. *J. Am. Chem. Soc.* **1977**, *95*, 8339–8341. (b) Noyori, R.; Tomino, I.; Yamada, M.; Nishizawa, M. *J. Am. Chem. Soc.* **1984**, *106*, 6717–6725. (c) Midland, M. M.; Tramontano, A.; Kazubski, A.; Graham, R. S.; Tsai, D. J. S.; Cardin, D. B. *Tetrahedron* **1984**, *40*, 1371–1380. (d) Ramachandran, P. V.; Teodorovic, A. V.; Rangaihenvi, M. V.; Brown, H. C. *J. Org. Chem.* **1992**, *57*, 2379–2386. (e) Bach, J.; Berenguer, R.; Garcia, J.; Loscertales, T.; Vilarrasa, J. *J. Org. Chem.* **1996**, *61*, 9021–9025.
 6. (a) Corey, E. J.; Helal, C. *J. Tetrahedron Lett.* **1995**, *36*, 9153–9156. (b) Helal, C. J.; Magriotsis, P. A.; Corey, E. J. *J. Am. Chem. Soc.* **1996**, *118*, 10938–10939. (c) Parker, K. A.; Ledebotter, M. W. *J. Org. Chem.* **1996**, *61*, 3214–3217.
 7. (a) Bradshaw, C. W.; Hummel, W.; Wong, C. H. *J. Org. Chem.* **1992**, *57*, 1532–1536. (b) Ansari, M. H.; Kusumoto, T.; Hiyama, T. *Tetrahedron Lett.* **1993**, *34*, 8271–8274. (c) Burgess, K.; Jennings, L. D. *J. Am. Chem. Soc.* **1991**, *113*, 6129–6139. (d) Jeromin, G. E.; Scheidt, A. *Tetrahedron Lett.* **1991**, *32*, 7021–7024. (d) Frantz, D. E.; Fässler, R.; Carreira, E. M. *J. Am. Chem. Soc.* **2000**, *122*, 1806–1807.
 8. Ishihara, K.; Mori, A.; Arai, I.; Yamamoto, H. *Tetrahedron Lett.* **1986**, *26*, 983–986.
 9. Mikami, K.; Yoshida, A.; Matsumoto, Y. *Tetrahedron Lett.* **1996**, *37*, 8515–8518.
 10. Singer, R. A.; Shepard, M. S.; Carreira, E. M. *Tetrahedron* **1998**, *54*, 7025–7032.
 11. (a) Brown, H. C.; Jadhav, P. K. *J. Am. Chem. Soc.* **1983**, *105*, 2092–2093. (b) Smith III, A. B.; Ott, G. R. *J. Am. Chem. Soc.* **1996**, *118*, 13095–13096.
 12. (a) Hafner, A.; Duthaler, R. O.; Marti, R.; Rihs, G.; Rothe-Streit, P.; Schwarzenbach, F. *J. Am. Chem. Soc.* **1992**, *114*, 2321–2336. (b) Riediker, M.; Duthaler, R. O. *Angew. Chem., Int. Ed. Engl.* **1989**, *28*, 494–495.
 13. (a) Sharma, A.; Chattopadhyay, S. *J. Org. Chem.* **1998**, *63*, 6128–6131. (b) Journet, M.; Cai, D.; DiMichele, L. M.; Larsen, R. D. *Tetrahedron Lett.* **1998**, *39*, 6427–6428.
 14. The enantiomeric excess of propargylic alcohol **2a–h** was determined by HPLC analysis (Waters 590 chromatograph coupled with a Varian UV detector, $\lambda=220$ nm). For (−)-**2a** and (+)-**2b**: Chiralpak-AD₂ column, (hexane/isopropanol: 99.2/0.8); For (−)-**2c**: Chiralpak-AD₂ column, (hexane/isopropanol: 93/7); For (−)-**2d** and (+)-**2e**: Chiralpak-OD-H column, (hexane/isopropanol: 99/1); For (−)-**2f**: Chiralcel-OD-H column, (hexane/isopropanol: 95/5); For (−)-**2g**: Chiralcel-OD-H column, (hexane/isopropanol: 90/10); For (−)-**2h**: Chiralcel-OD-H column, (hexane/ethanol: 94/6).
 15. Trost, B. M.; Belletire, J. L.; Godleski, S.; McDougal, P. G.; Balkovec, J. M.; Baldwin, J. J.; Christy, M. E.; Ponticello, G. S.; Varga, S. L.; Springer, J. P. *J. Org. Chem.* **1986**, *51*, 2370–2374.
 16. Lee, E.; Lee, Y. R.; Moon, B.; Kwon, O.; Shim, M. S.; Yun, J. S. *J. Org. Chem.* **1994**, *59*, 1444–1456.